

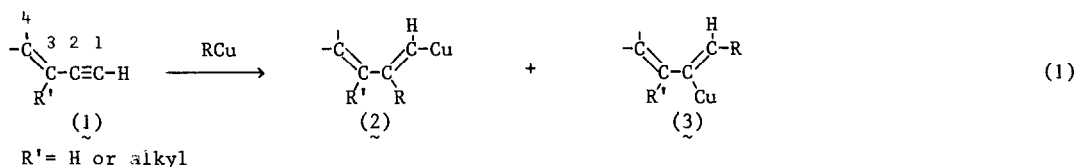
AN ORGANOCOPPER MEDIATED SYNTHESIS OF (*E*)-1,4-DI-TERT-BUTYLBUTATRIENE.

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Summary In the paper a highly stereoselective route to *trans*-1,4-di-tert-butylbutatriene via *t*-BuCu induced 1,3-substitution in (*Z*)-*t*-BuCH=C(Br)-C≡CH is described.

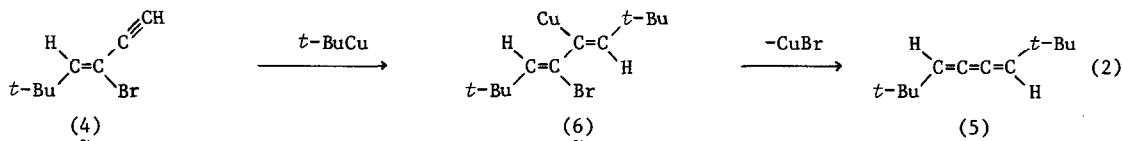
In the literature a number of reports concern the reaction of conjugated enynes with organocopper compounds. The available data show that the reaction gives adducts 2 (see eq.1) when the group R is small, and mixtures of adducts 2 and 3 when the group R is bulky.^{1,2} Both adducts 2 and 3 arise by addition of the copper reagent to the triple bond.



It is noteworthy in this connection that in one case addition of a copper reagent to the double bond of a conjugated enyne has been observed.³

At the moment that this study was initiated it was not known how copper reagents react with conjugated enynes in which on C(3) instead of R' a good leaving group (*e.g.* Br) is present. If the reaction of copper reagents with such functionalised enynes should give adducts of type 3, a subsequent elimination of CuX from these adducts should give cumulated trienes.

To investigate this potential route to butatriene derivatives it was decided to treat a suitably substituted 3-bromo-3-alken-1-yne with an organocopper compound. 3-Bromo-3-alken-1-yne 4 are readily available by reaction of 1,3-diyne with alkylsilver compounds and subsequent reaction of the produced adducts with *N*-bromosuccinimide.⁴ For our purpose we prepared the 3-bromo-3-alken-1-yne 4 (see eq.2). We were very pleased to find that treatment of 4 with an equimolar amount of *t*-BuCu during 15 min at -50°C in tetrahydrofuran gave in a regiospecific reaction pure triene 5 (see eq.3) in excellent yield (95%). ¹H NMR analysis (see below) showed that this regiospecific reaction was also highly stereoselective, if not stereospecific. Thus starting from 4 (*E/Z*=7/93) compound 5 has been obtained as a mixture consisting of 93% *trans* and 7% *cis* isomer. This stereochemical course is conceivable by assuming that the overall substitution follows the two-step reaction sequence depicted in eq.2:



The first step of the reaction is proposed to be a *cis* addition of *t*-BuCu to the triple bond in 4. This proposal is quite reasonable in view of the well-known *cis* addition of organocopper reagents to other 1-alkynes.^{1,2,5} The second reaction step in eq.2 must then involve an *anti* elimination of CuBr from adduct 6. Unfortunately, attempts to intercept adduct 6 by protolysis at -60°C have failed thus far, so that a synchronous 1,3-substitution path cannot be excluded.

The *trans* geometry proposed for 5 is based on the comparison of the ^1H NMR data recorded for this compound with those found for *cis*-1,4-di-*tert*-butylbutatriene. The *cis* butatrienyl compound is readily available from ethynyl methyl sulfide and *t*-BuCu.^{6,7} Its ^1H NMR spectrum (CCl_4 ; $\delta(\text{Me}_4\text{Si}) = 0$ ppm) shows two sharp singlets, one at δ 5.44 (=CH) and the other at δ 1.11 (*t*-Bu). Our compound 5 on the other hand shows sharp singlets at δ 5.48 (=CH) and δ 1.09 (*t*-Bu). A mixture of both compounds shows the expected four signals in the ^1H NMR spectrum. The overall stereochemistry of the conversion of 4 into 5 is therefore *anti*. To our best knowledge this is the first example of an organocopper induced 1,3-substitution in a propargylic compound in which the carbon bearing the leaving group is sp^2 -hybridized. Also organocopper induced 1,3-substitution reactions in propargylic esters in which the propargylic carbon is sp^3 -hybridized generally proceed *anti*.⁸

Currently the scope of the presented reaction principle is investigated.

References

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4. H. Kleijn, M. Tigchelaar, J. Meijer, and P. Vermeer, *Recl. Trav. Chim. Pays-Bas*, in press.
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7. That this reaction indeed gives *cis*-butatriene derivatives has been established by X-ray analysis of another butatriene compound which was prepared *via* the same method: W.G.M. van den Hoek, J. Kroon, H. Kleijn, H. Westmijze and P. Vermeer, and H.J.T. Bos, *J. Chem. Soc., Perkin II*, **1979**, 4231.
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